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***AP Chemistry
(Zumdahl) Videos***

***Zumdahl Answer
key, Chapter 13,
Question 39 (9th
edition), Question
45 (10th edition)*** The
equilibrium constant is
0.0900 at 258C for the

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Answers

reaction $\text{H}_2\text{O} + \text{Cl}_2\text{O}$

$\rightleftharpoons 2\text{HOCl}$ For

which of the following
sets of conditions ...

Balancing Chemical Equations Practice Problems

Equation
balancing will make
sense! Here, we will do
a bunch of practice
problems for balancing
chemical equations.
We'll see ...

DNA vs RNA

(Updated) Why is RNA

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just as cool as DNA?
Join the Amoeba
Sisters as they

compare and contrast
RNA with DNA and
learn why DNA ...

***Chapter 4 (Types of
Chemical Reactions
and Solution***

Stoichiometry) - Part

1 Major topics:

solution vocabulary,
solvation, strong vs.
weak electrolytes,
molarity, & dilution
vocabulary,

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techniques ...

Edition Answers

**Acid-Base Reactions
in Solution: Crash
Course Chemistry #8**

Last week, Hank talked about how stuff mixes together in solutions. Today, and for the next few weeks, he will talk about the ...

**Chapter 11
(Properties of
Solutions)** Major
topics: **solution**
concentration

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calculations (molarity, percent by mass, mole fraction), steps of **solution** formation, heat of ...

Zumdahl Answer key, Chapter 13, Question 41 (9th edition), 47 (10th

edition At 9008C, K_p 5 1.04 for the reaction $\text{CaCO}_3 \rightleftharpoons \text{CaO} + \text{CO}_2$ At a low temperature, dry ice (solid CO_2), calcium oxide, and calcium ...

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Photosynthesis:

Crash Course

Biology #8 Hank

explains the extremely complex series of reactions whereby plants feed themselves on sunlight, carbon dioxide and water ...

Zumdahl Answer

key, Chapter 13,

Question 40 (9th

edition) The

equilibrium constant

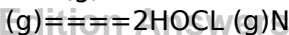
K_p is 0.0900 10³ at

25C for the reaction

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$\text{N}_2(\text{g})$ For which of
the ...

***Nuclear Chemistry:
Crash Course***

Chemistry #38 You
can directly support
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shcourse](http://www.subbable.com/crashcourse) Subscribe for
as little as \$0 to keep
up with ...

***Describing Reactions
in Solutions***

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Describing Reactions in
Solutions Considering
the types of equations
used to represent
reactions in **solutions**:
1- The ...

***Mitosis: The
Amazing Cell
Process that Uses
Division to Multiply!
(Updated)*** Updated
Mitosis Video. The
Amoeba Sisters walk
you through the reason
for mitosis with
mnemonics for

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prophase, metaphase
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Biomolecules

(Updated) This video, as stated in the description, focuses on general functions of biomolecules. The biomolecules: carbs, lipids ...

Polar & Non-Polar Molecules: Crash Course Chemistry #23 PLEASE WATCH WITH ANNOTATIONS

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Edition Answers

ON! SOME
INACCURACIES IN
GRAPHICS ARE NOTED
AND CORRECTED IN ...

***DNA Structure and
Replication: Crash
Course Biology #10***

Hank introduces us to
that wondrous
molecule

deoxyribonucleic acid -
also known as DNA -
and explains how it
replicates itself ...

ATP & Respiration:

Page 14/24

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Crash Course

Biology #7 In which Hank does some push ups for science and describes the "economy" of cellular respiration and the various processes ...

**Atomic Hook-Ups -
Types of Chemical
Bonds: Crash Course
Chemistry #22** Atoms are a lot like us - we call their relationships "bonds," and there are many different types.

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**Solubility Chemistry
- Solute Solvent &
Solution, Weak
Electrolytes Strong
Electrolytes &
Nonelectro** This

chemistry video provides a basic introduction into solubility and how compounds dissolve in water. It discusses how ...

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Mutations (Updated)

Join the Amoeba

Sisters as they explain gene and chromosome mutations, and explore the significance of these changes. This ...

How to Get Answers for Any Homework or Test

I am going back to school so I can have my degree once and for all. I work about 50-60 hours a week while going to school, so I

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Types of Chemical

Reactions We'll learn

about the five major

types of **chemical**

reactions: synthesis,

decomposition,

synthesis, single

replacement (also

called ...

The Ideal Gas Law:

Crash Course

Chemistry #12 Gases

are everywhere, and

this is good news and

bad news for chemists.

The good news: when

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they are behaving

themselves, ...

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three cornered

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